Hvordan foredle biogass og redde verden i samme slengen



(English: How to purify biogas and save the world!)

Presenter: Fred Martin Kaaby, MSc. Organic Chemistry

Target audience: Technicians, CEOs, decision makers, engineers

Todays agenda

01	Who are we?
02	Biogas and biomethane
03	Impurities in biogas and their adverse effects
04	EU requirements for quality of biomethane
05	Online monitoring of methane purity
06	Periodic measurements of gas quality
07	Purification methods
08	Summary and conclusions



Institute For Energy Technology's vision: Internationally leading research institute

Turnover:

1

BNOK

Annual publications:

120



1948: IFA



1980: IFE

Employees:

300



14.000

Annual visitors

Advanced laboratories:

24



Nationalities: 32

Researchers: 218

PhDs: 105

Centres for renewable

energy:

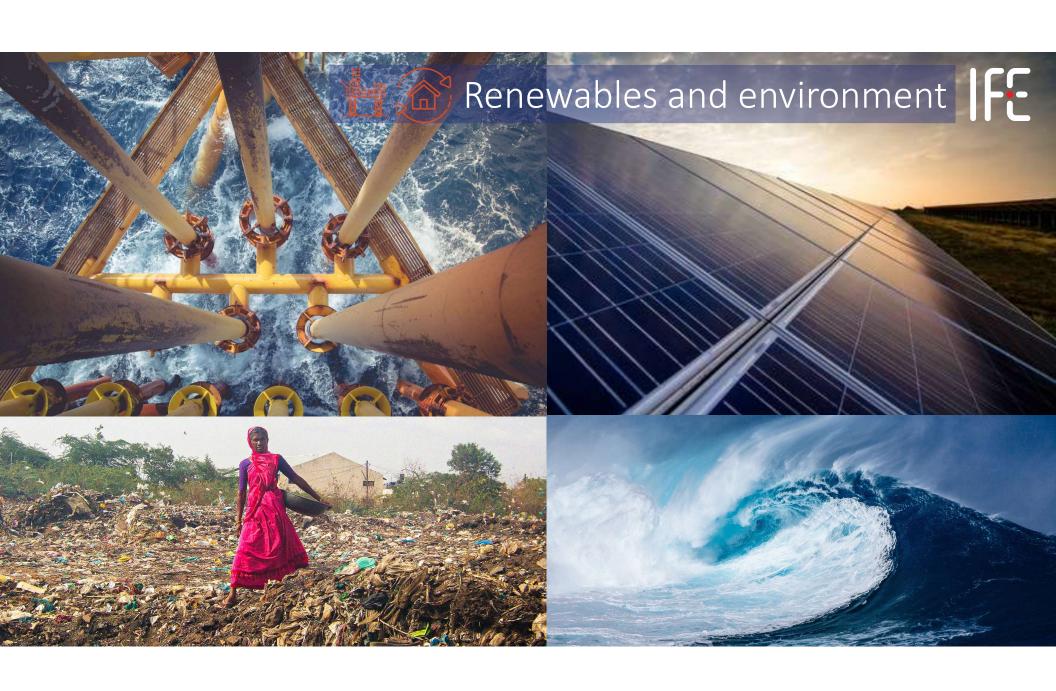
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International projects:

>30%





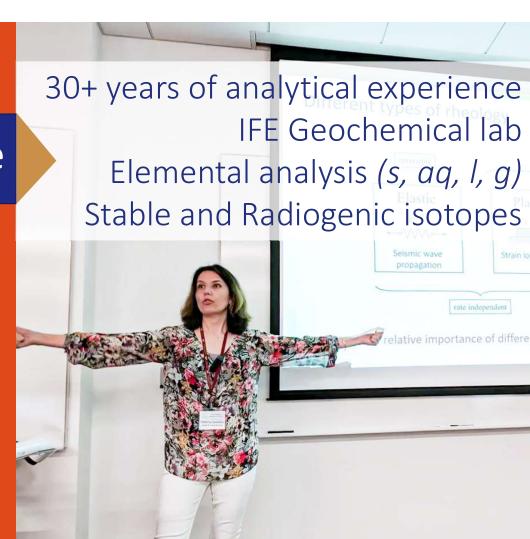




Gas composition analysis

Periodic measurements

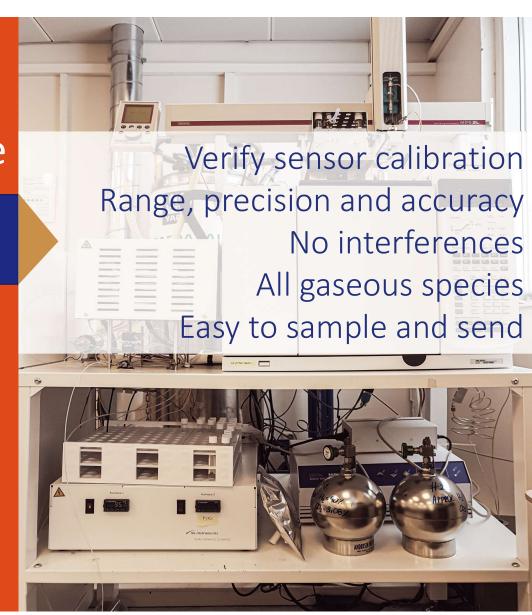
Training



Gas composition analysis

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Gas composition analysis

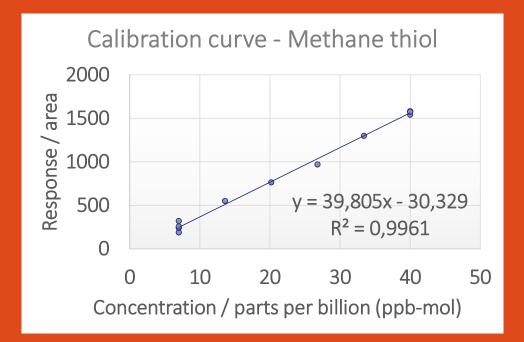
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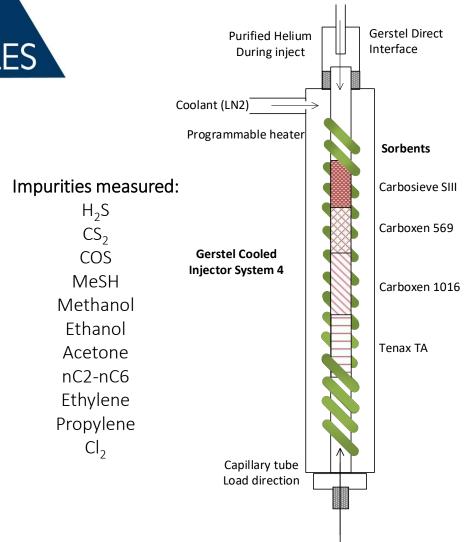
Training





Quantification of hydrogen gas impurities according to EN17124



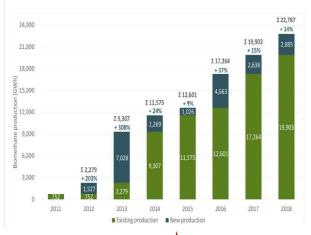


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IFE

Biogas is climate neutral



EBA Biogas potential in Europe: 1170 TWh (2050)

Biogas



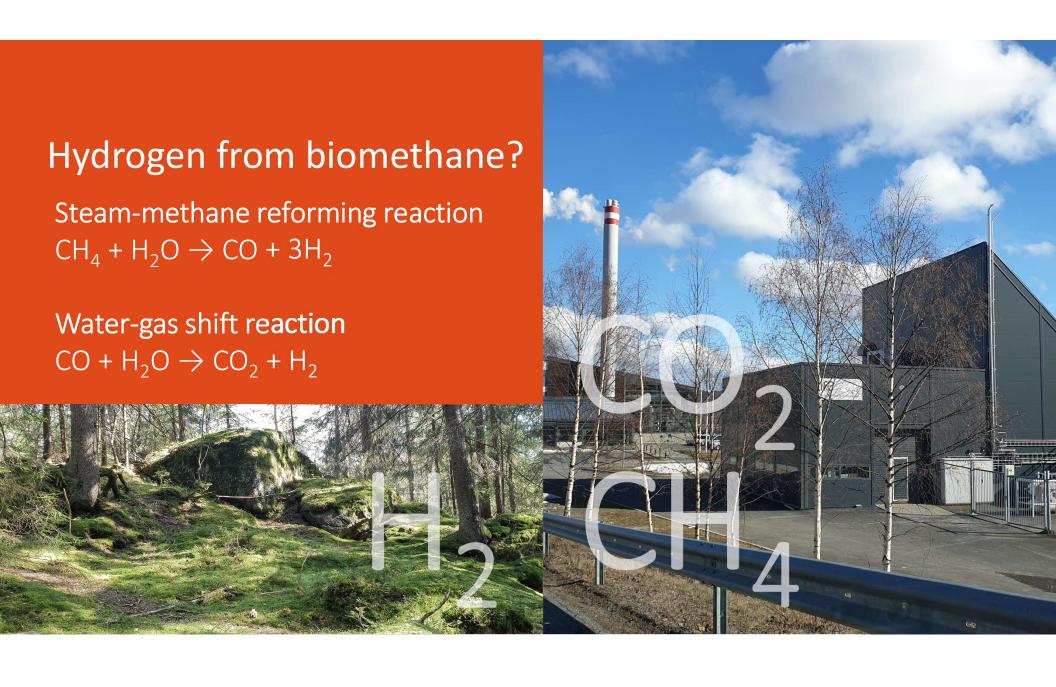












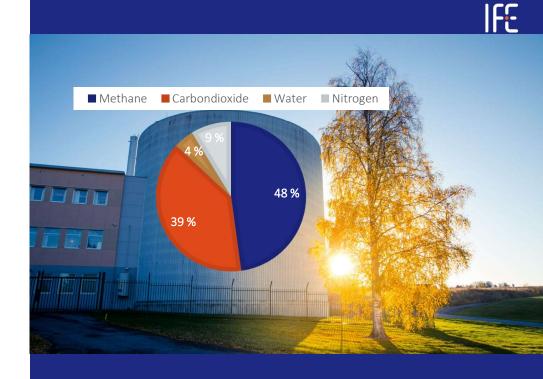
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Biogas (biowaste) content

Methane (CH_4) 55–70% Carbon dioxide (CO_2) 30–45%

 $\begin{array}{lll} \mbox{Nitrogen (N_2)} & < 15\% \\ \mbox{Oxygen (O_2)} & < 3\% \\ \mbox{Water (H_2O)} & < 5\% \\ \mbox{Hydrogen sulfide (H_2S)} & < 1\% \\ \mbox{Ammonia (NH_3)} & < 0.1\% \\ \mbox{Siloxanes} & < 0.1\% \\ \mbox{Terpenes} & < 0.1\% \end{array}$

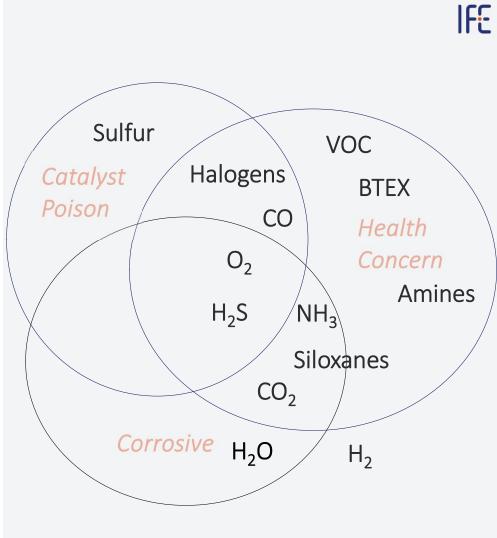


Volatile organic compounds (VOC)
All organic compounds with high vapor pressure and low water solubility.

Biogas impurities and risks

Methane is flammable!





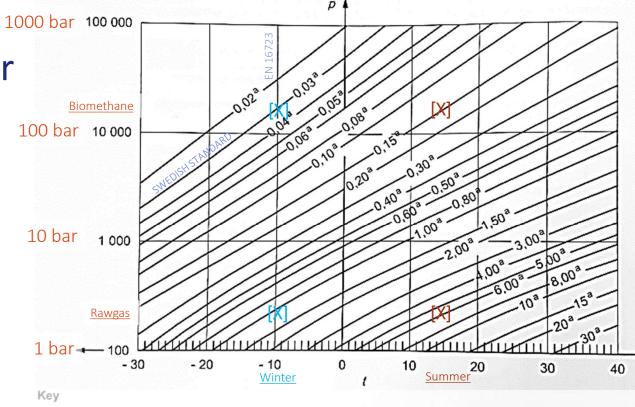


Water saturation limits – dew point (g/m³)

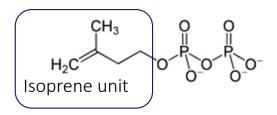
How to handle water

The art of manipulation of P and T



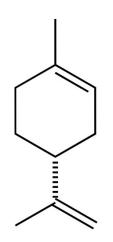


- temperature, °C
- absolute pressure, kPa
- Water content of natural gases in g/m³ (0 °C, 101,325 kPa, dry)

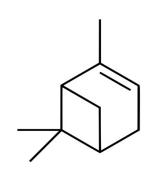


Terpenes and terpenoids

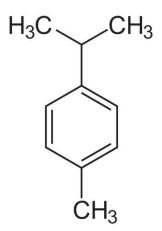
Built by nature from isoprene units.



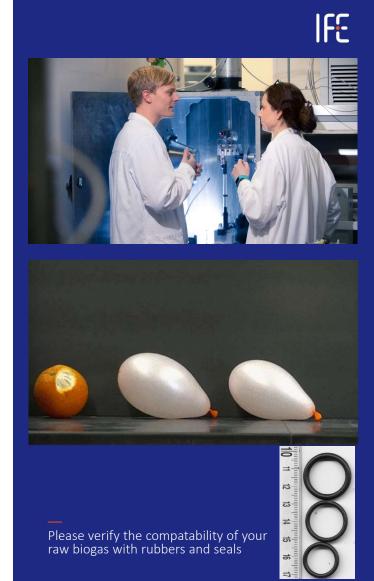
Limonene gives lemon its smell



Pinene is main constituent in terpentine

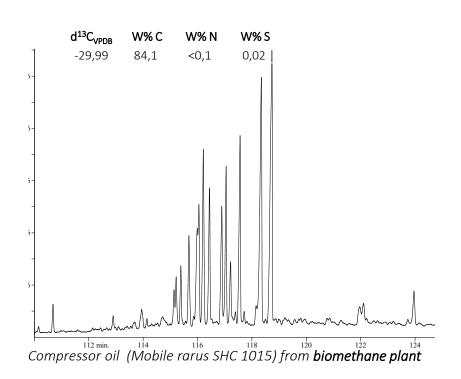


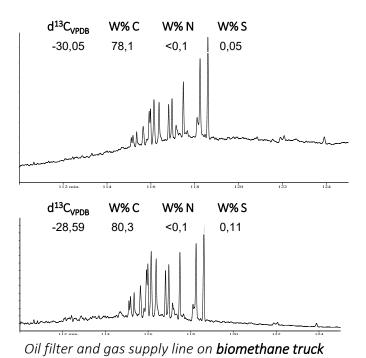
Cymene is a naturally occurring aromatic organic compound





The challenge with compressor oil





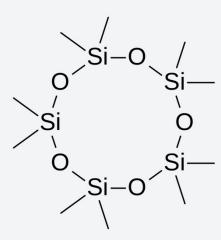
IFE

Siloxanes



Combustion of siloxanes creates silicon dioxide (glas).





D5 – Decamethylcyclopentasiloxane - Is listed in The Candidate List of Substances of Very High Concern (SVHC).



ICP-MS analysis of scales on underside of exhaust vent and injection vent from A damaged motor inside a **biomethane truck**

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Biomethane for Automotive fuel (EN 16723-2)

Total silicon (siloxanes) b

Hydrogen

Hydrocarbons (including terpenes)

Oxygen

Total sulfur

Methane

Compressor oil

Dust

Amines (including NH₃)

Water

 $< 0.5 \text{ mg/m}^3$

< 2 %

< -2 °C dew point

< 1 %

 $< 5 \text{ mg/m}^3$

> 65 % (High grade 80)

de deminisa

de deminisa

 $< 10 \text{ mg/m}^3$

< -10 °C dew point



b) Some gas turbines manufacturers have set $< 0.1 \text{ mg/m}^3 \text{ limits}$.



The biomethane monitoring caveat:

«Of all the requirements, only methane, CO2, hydrogen sulfide, oxygen is monitored real-time.»

Hydrogen specification for PEM fuel cells (EN 17124:2018)

Nitrogen	< 300 µmol/mol
Methane	< 100
Water	< 5
Oxygen	< 5
Carbon dioxide	< 2
Total hydrocarbons (except CH4)	< 2
Carbon monoxide	< 0.2
Formic acid	< 0.2
Ammonia	< 0.1
Halogenated compounds*	< 0.05
Formaldehyde	< 0.2
Total CO, formic acid, formaldehyde	< 0.2
Total sulphur compounds	< 0.004
Particulate	< 1 mg/kg



Hydrogen production from biomethane?

Purification is even more essential!

Biomethane gas quality standards

	Natural gas network	Automotive fuel	Swedish standard	Suggested test methods
	EN 16723-1	EN 16723-2	SS 15 54 38:1999	
Total silicon (siloxanes) c	< 0.3 (pure) - 1 (diluted) mg/m ³	< 0.5 mg/m ³		EN 16017-1
Hydrogen		< 2 %		ISO 6974, 6975
Hydrocarbons		< -2 °C dew point		ISO 23874, 11150, 12148, 6974
Oxygen		< 1 %	< 1 %	ISO 6974, 6975
Total sulfur	< 20 mg/m³	< 5 mg/m ³	< 23 mg/m³	ISO 6326, 19739
Methane		> 65 % (High grade 80)	97 % ± 2	EN 16726
Compressor oil	de deminis ^a	de deminisª		ISO 8573-2
Dust	de deminis ^a	de deminisª		ISO 8573-4
Amines	< 10 mg/m³	$< 10 \text{ mg/m}^3$	< 20 mg/m³	VDI 2467 Blatt2: 1991-08
Water		< -10 °C dew point	< 32 mg/m³	ISO 10101, 6327, 11541
Chlorinated compounds	de deminis ^a			ISO 6142-1, HCl: EN 1911
Fluorinated compounds	de deminisª			NF X43-304, ISO 15713
Carbon monoxide	< 0.1 %			ISO 6974
Ammonia	< 10 mg/m³		< 20 mg/m³	NEN 2826, VDI 3496 Blatt 1:1982-04, NF X43-303
Wobbe index			43,9 - 47,3 MJ/m³	ISO 6976, 6974
Carbon dioxide + Nitrogen			< 3-4 %	ISO 6974
Motoroktantall (MON) b			> 130	ISO 15403

a) de deminis - Does not render the fuel unacceptable for use in end user applications

Sampling according to ISO 10715, 16726

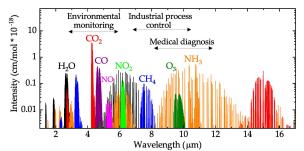
b) MON = 137,78*metan+181,233*CO2+26,994*N2

c) Some gas turbines manufacturers have set $< 0.1 \text{ mg/m}^3 \text{ limits}$.

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Nondispersive infrared spectra of common gases

Common online sensors and their limitations

Wikimedia Commons@Daniel Popa and Florin Udrea

Oxygen

Mostly electrochemical, finite lifespan (2-3 years).

Calibration check: 3 months.

Hydrogen sulfide

Mostly electrochemical, finite lifespan (1-2 years). Cross-sensitive to hydrogen.

Calibration check:
3 months. Requires personell handling of toxic gas. Gas mixtures have limited «shelf-time».

<u>Carbondioxide/Methane</u>

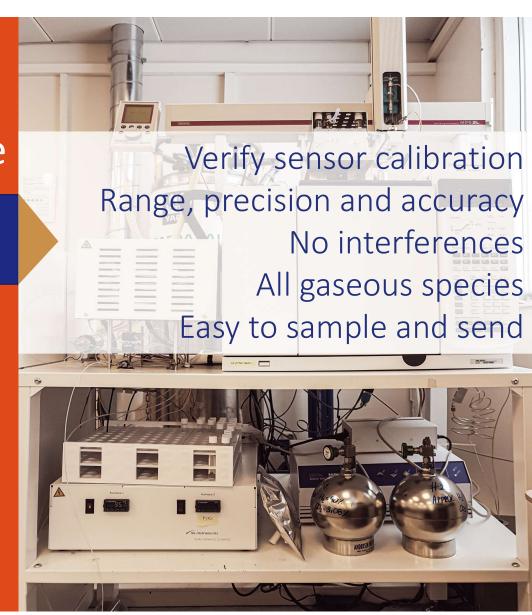
Nondispersive infrared sensors are bulky and expensive, but robust!

Calibration check: 6 months

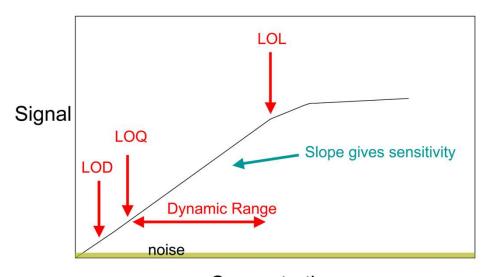
Gas composition analysis

Periodic measurements

Training



Frequent calibration is key for accuracy in online monitoring



Concentration



Five habits for a good calibration of sensors

- Learn the basics on handling pressurised gas before continuing.
- 2. Always include background (zero) points
- 3. Minimum two calibration points excluding background.
- 4. Calibrate with certified standards within expected «shelf-time».
- 5. Purge thoroughly the gas to wash out background air and saturate lines.

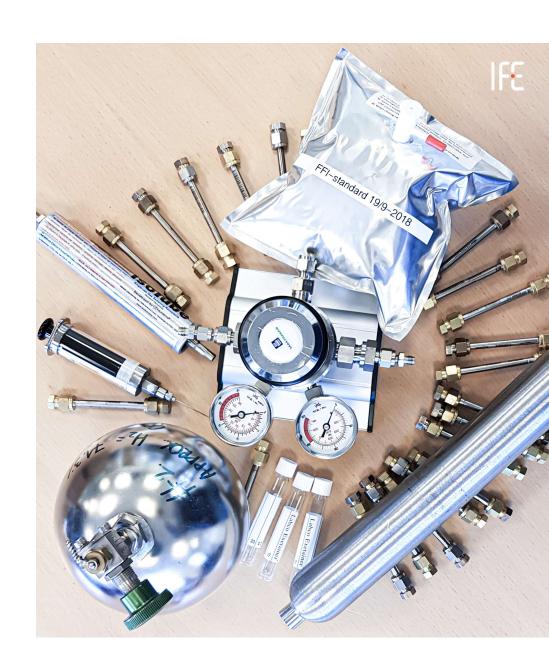
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Periodic measurements

Twice/year and after feedstock/process changes according to EN16723





Gas composition analysis

Periodic measurements

Training



Gas composition analysis

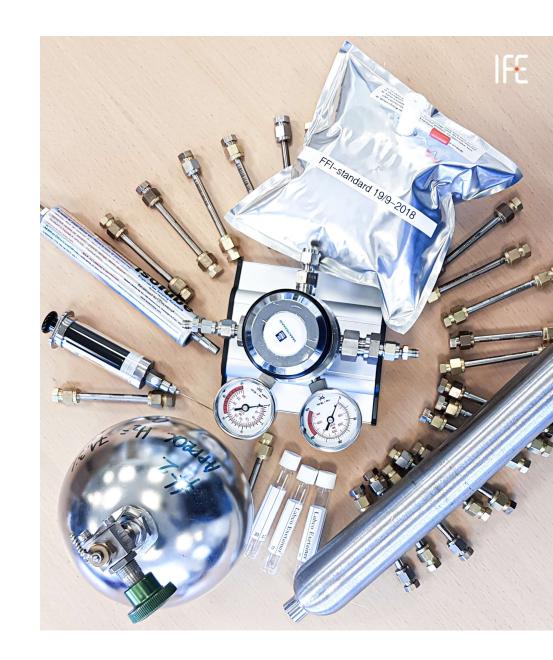
Periodic measurements

Training



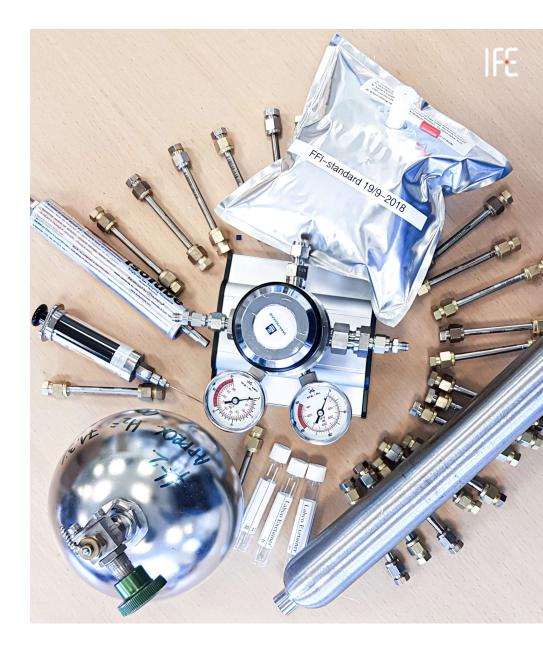
Five good habits on sampling gas with bags

- 1. Make sure you are using the correct type of gas bag for the purpose of the sampling
- 2. Remember even empty gas bags contains air.
- 3. Never reuse a gas bag.
- 4. Always use soft-tubing which are inert to your gas.
- 5. Purge well and thoroughly all lines before sampling.



Five good habits on sampling gas with TD-tubes

- 1. Always sample in the middle of the flow! VOC's creeps on the tubing walls.
- 2. Start sampling on a «training tube» to assert flow readings.
- 3. Split the flow, don't restrict it, to get a accurate sample within the limits of your tube.
- 4. Inertness of tubing and equipment is very important in VOC-analysis.
- 5. Sample several tubes with different sampling-times. Example: <15s, 30s, 1', 2'> at 80-100 mL/min

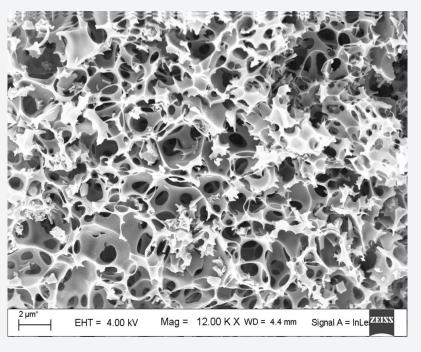


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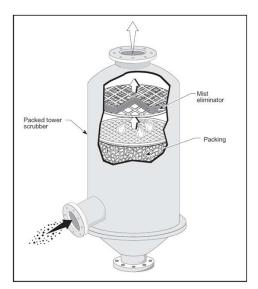
Absorbents, adsorbents / chemisorption vs physisorption



 $2 H_2S(g) + 3 O_2(g) \rightarrow 2 SO_2(g) + 2 H_2O(I)$

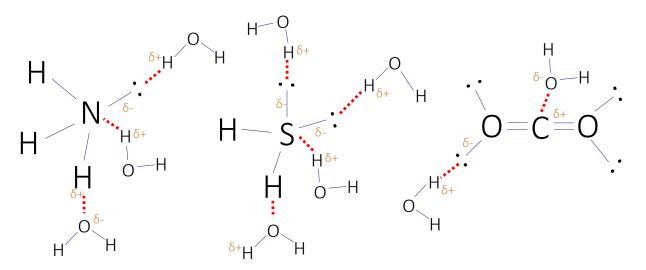
Vikimedia Commons - CC BY 4.0 SomnathWiki001





Scrub the gas!

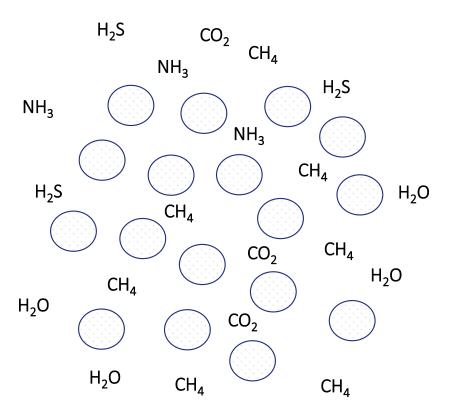
Use the power of hydrogen-bonds





Pressure swing adsorption

Pressure manipulation increases adsorption.





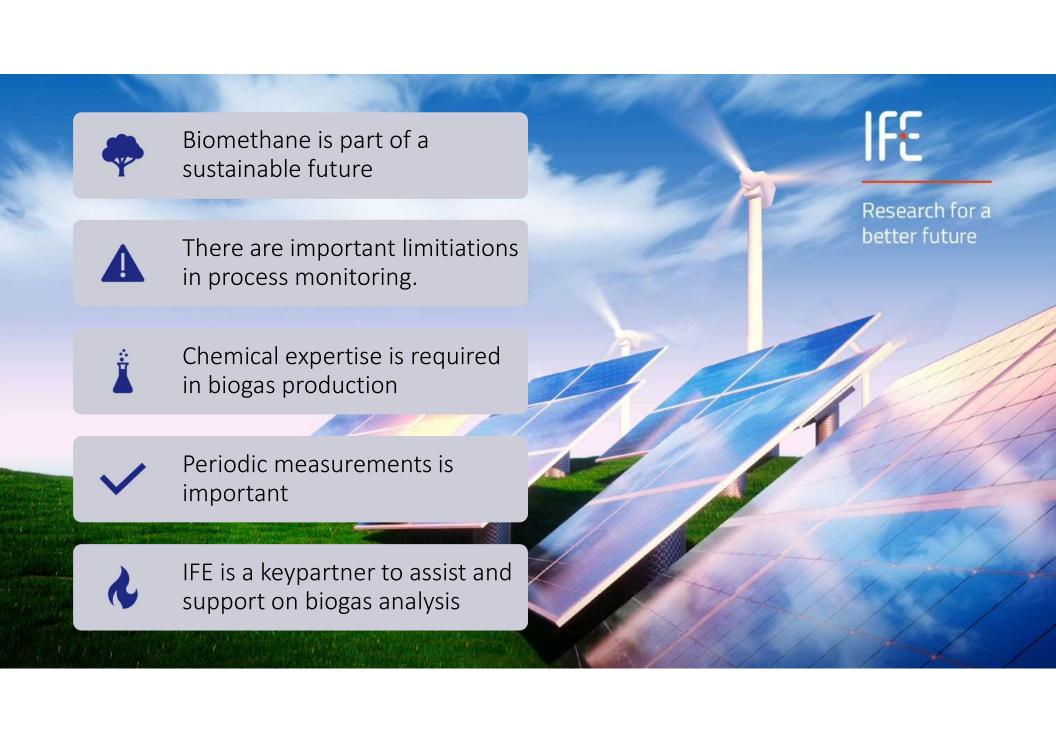




There are many other upgrading technologies not mentioned here. Example: Membrane-technology have come a long way.

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Thank you for your attention!

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